



Eastern University, Sri Lanka Second Year Second Semester Examination in Science-2010/2011 (April/ May 2012) CH 205 Boron Chemistry and Silicates

(Proper and Repeat)

Answer all questions	Time: 01	hour
1. (a) Briefly explain the Wade's rule		
		(10 Marks)
(b) Place the each of the following molecules/ions in the correct	ct class.	
i. B ₅ H ₁₂ ⁻ ii. B ₃ C ₂ H ₅ iii. B ₅ H ₁₁		
	1, 4	(30 Marks)
(c) Derive the possible structure (s) of the following boron compounds using Wade's rule and		
draw the structure by indicating all the bonds.	5	
i. B ₅ H ₁₁ ii. B ₆ H ₁₀		
· · · · · · · · · · · · · · · · · · ·)	(20 Marks)
(d) Derive the different 'styx' numbers possible for B_5H_{11}	•	
(i) Choose the most likely 'styx' number		
(ii) Draw the possible structure of B_5H_{11}		
		(20 Marks)
(e) Show by means of equations how the following transformations could be done.		



(20 Marks)

Contd...

2. (a) Write down the end product (s) of the following reactions.

i.
$$B_2H_6$$
 LiR
ii. B_2H_6 (CH_3)3N
iii. B_2H_6 Cl_2, 25°C

(b) Classify the silicate minerals by giving general formulas and one example.

(35 Marks)

(c) Describe the double chain silicates using suitable examples.

(50 Marks)