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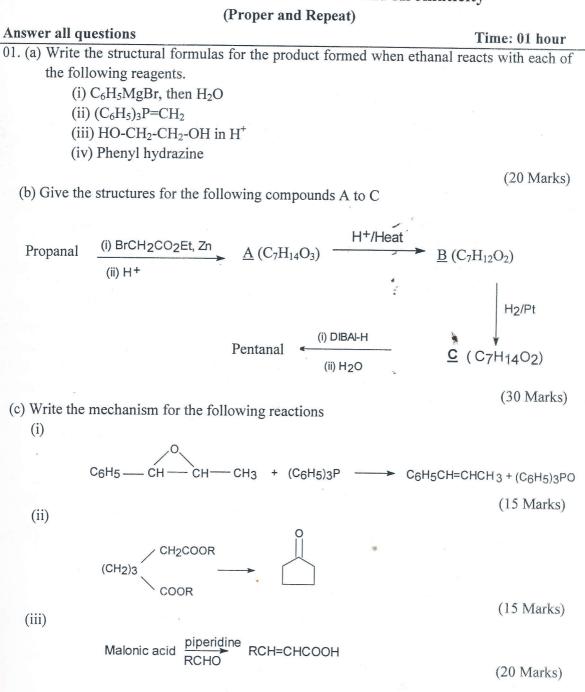
Eastern University, Sri Lanka

Second Year First Semester Examination in Science-2008/2009

(May/July 2012)

External Degree

EXTCH 204 Reaction Mechanisms and Aromaticity

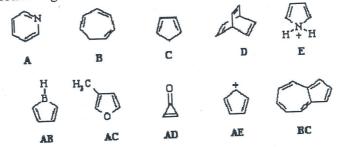


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CH 204 8/9 @

02. (a) (I) State Huckel's rule for aromaticity

(II) Use the following list of compounds to answer the questions below:



Choose the compound that is best described as the following;

(i) A neutral, 4 π -electron, anti-aromatic system.

(ii) A 6 π -electron, aromatic system.

(iii) A non-aromatic, conjugated 6 π -electron system.

(iv) A non conjugated hydrocarbon.

(v) Non-aromatic as drawn, but if H- were removed would give an aromatic cation.

(vi) Non-aromatic as drawn, but has an important resonance structure.

(vii) Non-aromatic as drawn but has an aromatic conjugate base.

(viii) An aromatic heterocyclic that can act as a diene in a Diels-Alder reaction.

(40 Marks)

(b) Use the polygon and circle method to outline the π molecular orbitals of cyclopentadienyl anion and hence explain whether cyclopentadienyl anion is aromatic or not. (20 Marks)

(c) Identify the product (with the correct stereochemistry where possible) in each of the following reaction.

