

EASTERN UNIVERSITY, SRI LANKA (2016) THIRD EXAMINATION IN SCIENCE-2012/2013 (MAY/June' 2016) FIRST SEMESTER (2017 2017

CH 302 Heterocyclic Chemistry and Molecular Rearrangement Reactions (Proper and Repeat)

Answer all questions

Time Allowed: One hour

- 1. a) The compound A was treated with compound B and then the formed product (Q) was treated with acid to get C.
 - i) Draw the structure of the product Q and R.
 - ii) Suggest a plausible mechanism for the reaction from Q to R.



(20 marks)

b) Identify the product (P) and write a suitable mechanism for the reaction.



Contd...

- c) When compound A was treated with hydroxylamine it gave two compounds P After separation, these products P (major) and Q were further treated separate sulphuric acid to get a compound R and S respectively. The rate of reaction formation of R was found to be faster than that of S.
 - i. Predict the two products P and Q
 - ii. Write the mechanism of the reaction P to S.
 - iii. Why is the compound \mathbf{P} obtained as the major product?



d) Predict the product(s) formed in the following reactions.



2. a) Suggest plausible mechanisms for the following reactions.



Cont

(30 marks



(30 marks)

b) Explain the following observations.

i. Elecrophilic substitution reactions of pyrrole furans and thiophenes occur predominantly at α position than at β .

(20 marks)

(30 marks)

iii. Quinoline is much more reactive towards electrophilic substitution than pyridine. (20 marks)

c) Suggest plausible mechanisms for the following reactions.

