

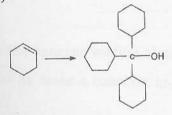
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## EASTERN UNIVERSITY, SRI LANKA SECOND EXAMINATION IN SCIENCE – 2003/2004 SECOND SEMESTER (June/July-2005) CH 205 BORON CHEMISTRY AND SILICATES

Answer All Questions.

Time: 01 Hour

- 1. a) State Wade's rule.
  - b) Discuss the type of bonding and structure in each of the following Boron compounds, using Wade's rule.
    1) C<sub>2</sub>B<sub>3</sub>H<sub>5</sub>
    2) [B<sub>4</sub>H<sub>9</sub>]
  - c) Derive "styx" number for B<sub>4</sub>H<sub>10</sub>, indicating the equations and rules clearly. And draw the most possible schematic diagram corresponding "styx" number.
  - d) How can the following transformation be effected through Boron compounds intermediate(s)?



- 2. a) (i) List the structural types of silicates.
  - (ii) Classify the following silicates into the different structure types
    - 1. Ca<sub>2</sub>Mg<sub>5</sub>(Si<sub>4</sub>O<sub>11</sub>)<sub>2</sub> (Tremolite)
    - 2. Be<sub>3</sub>Al<sub>2</sub>Si<sub>6</sub>O<sub>18</sub> (Beryl)
    - 3. Mg<sub>3</sub>(OH)<sub>2</sub>Si<sub>4</sub>O<sub>10</sub> (Talc)
    - 4. Be<sub>2</sub>SiO<sub>4</sub> (Phenacite)
  - b) Briefly describe the properties of the followings
    - (i) Ortho silicates
    - (ii) Zeolites

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