

EASETRN UNIVERSITY, SRI LANKA

SECOND EXAMINATION IN SCIENCE - REPEAT

SECOND SEMESTER 2004/2005 (OCTOBER 2006)

CH 205 BORON CHEMISTRY AND SILICATES

Time allowed: ONE Hour

Candidate must NOT start writing their answers until told to do so

CH 205 BORON CHEMISTRY AND SILICATES 2004/2005

- 1. (a) Place each of the following molecules/ionsⁱ in the correct structural class
 - (i) B_3H_8 (ii) B_5H_{12} (iii) B_3H_9 (iv) $B_3C_2H_7$ (20 marks)
 - (b) Derive a possible structure for the following molecules/ions and draw the diagram by indicating all the bonds

(i)
$$B_3H_8$$
 (ii) $B_3C_2H_7$ (40 marks)

- (c) (i) Derive the different 'styx' numbers possible for B_4H_{10} (24 marks)
 - (ii) Choose the most likely 'styx' number obtained in c (i) above and draw the structure

(16 marks)

2. (a) Show by means of equations how the following transformations could be effected via organometallic intermediates

(i) $CH_3CH = CH_2$		CH ₃ CH ₂ CH ₃	(20 marks)
(ii) B ₃ N ₃ H ₆	>	[BNH(NHR)] ₃	(20 marks)

(b) Write balanced equations to show the reactions of (BNHCl)₃ with MeMgBr, H₂O and LiAlH₄. (10 marks)

- (c) Silicate minerals are classified according to the linkage pattern of SiO₄⁴⁻ tetrahedral units. List the types of mineral by giving
 - (i) general formula

(ii) schematic diagrams (description is not necessary)

(iii) only one example for each

(50 marks)