

Time: 01 hour.

Answer ALL Questions



 Briefly describe the differences between single and poly crystalline materials. What is a space lattice? Demonstrate by suitable diagrams, how a crystal structure can be obtained from a space lattice and basis. A unit cell of a body centered crystal structure is shown below.



- (a) State whether this unit cell is primitive or not. Explain your answer.
- (b) Find the packing fraction of a body centered cubic structure.
- (c) If the primitive translational vectors of a body centered cube is given by:

$$b_1 = \frac{a}{2}(i+j-k)$$
, $b_2 = \frac{a}{2}(-i+j+k)$ and $b_3 = \frac{a}{2}(i-j+k)$

where $a_1 = ai$, $a_2 = aj$ and $a_3 = ak$ are the basis vectors of the unit cell and a is the lattice side, find the volume of the primitive unit cell.

(d) Hence or otherwise find the reciprocal lattice vectors of a body centered crystal structure.

2. Show that for any cubic lattice the separation of the planes corresponding to Miller indices (*hkl*) is given by:

$$d_{hkl} = \frac{a}{\sqrt{h^2 + k^2 + l^2}}$$

Where *a* is the lattice parameter.

Briefly describe the Bragg's diffraction in crystals and show that the Bragg condition for crystal diffraction on (hkl) planes is given by:

$$2d_{hkl}sin\theta_{hkl} = n\lambda,$$

Where the symbols have their usual meanings.

Determine the Bragg angles for the (111), (220), (311), and (400) reflections of Germanium which has a cubic structure with lattice parameter 5.65 A° using "Copper K_a" X-rays which has a wavelength $\lambda = 0.154$ nm.