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15 JAN 2009	A CONTRACT OF CONTRACT OF CONTRACT		
In University Sri Lanker E.	ASTERN UNIVERSITY, S	RI LANKA	
FIRST SEM	<b>IESTER THIRD EXAMIN</b>	ATION IN SCIENCE	
	SPECIAL - REPEA	The sub-substance the ST	
	2007/2008(Dec.2008	3)	
CH 305 ORGAN	NOMETALLIC CHEMIST	RY & NON – AQUEOUS	
	SOL VENTS		

Time: 01 Hour

Answer all questions

1. a) Indicate the monohapto, dihapto, trihapto, tetrahapto, pentahapto and bridging ligands present in the following compounds



b) Give the systematic names of the following organometallic compounds.

i)  $\left[ Cr(CO)_4(PR_3)_2 \right]$ 



Fe

- c) i) Arrange the following compounds in the order of increasing stretching frequency of the C-O bond. Account for your arrangement.
  CO, [V (CO)<sub>6</sub>]<sup>-</sup>, [Cr(CO)<sub>6</sub>], [Mn(CO)<sub>6</sub>]<sup>+</sup>
  - ii) A diamagnetic organometallic compound  $\underline{\mathbf{P}}$  having molecular formula:  $Co_2(CO)_8$  shows strong absorption at 2000 cm<sup>-1</sup> and 1805 cm<sup>-1</sup> in the region where CO stretching frequencies are observed. The <sup>13</sup>C nmr spectrum of  $\underline{\mathbf{P}}$ consist of two signals of relative intensity 1(singlet): 3(singlet). Deduce the structure of  $\underline{\mathbf{P}}$ .

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- a) What is meant by EAN rule? Indicate whether the following organometallic compounds obey Effective Atomic Number (EAN) rule or not. (Atomic 1. number: V = 23, Co = 27, Fe = 26, Cr = 24)
  - i)  $\begin{bmatrix} V(CO)_6 \end{bmatrix}$  ii)  $\begin{bmatrix} Co(CO)_3 NO \end{bmatrix}$
  - b) Give balanced chemical equations for the following reactions.
    - i. SbF5 in HF.
    - ii. H<sub>2</sub>SO<sub>4</sub> in C<sub>2</sub>H<sub>5</sub>OH.

c) Explain the following with appropriate reasons.

- i. Acetic acid behaves as a weak acid in aqueous solution but shows strong acidic property in liq. NH<sub>3</sub>.
- ii. Water acts as a leveling solvent for strong acids whereas liq. NH<sub>3</sub> act as differentiating solvent